

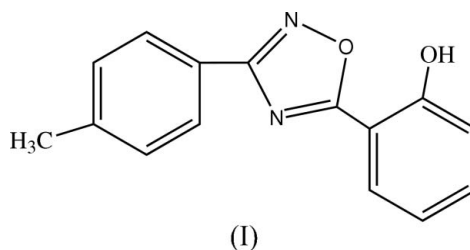
2-[3-(4-Methylphenyl)-1,2,4-oxadiazol-5-yl]phenol

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Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.008$ Å
 R factor = 0.063
 wR factor = 0.181
Data-to-parameter ratio = 7.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.In the approximately planar molecule of the title compound, $\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}_2$, which is a derivative of oxadiazole, an intramolecular $\text{O}-\text{H}\cdots\text{N}$ hydrogen bond helps to establish the molecular conformation.

Comment

As part of our ongoing structural studies of 1,2,4-oxadiazole derivatives (Ding *et al.*, 2006), we report here the synthesis and crystal structure of the title compound, (I) (Fig. 1).The dihedral angles between the $\text{N}2/\text{C}9/\text{O}1/\text{N}1/\text{C}8$ ring and its adjacent benzene rings are 3.9 (3) and 5.0 (3)° for the $\text{C}10$ and $\text{C}5$ rings, respectively. An intramolecular $\text{O}-\text{H}\cdots\text{N}$ hydrogen bond (Table 1) helps to establish the molecular conformation of (I).

Experimental

4-Methylbenzoic acid amidoxime (50 mmol) and methyl salicylate (100 mmol) were dissolved in anhydrous ethanol (150 ml) mixed with sodium (2.3 g) and heated for 3×25 min (with a 5 min break each time) at 400 W in a microwave. The reaction mixture was then concentrated by evaporation *in vacuo* to about one-third of its volume and the residue was mixed with water (20 ml). Whilst cooling the mixture, the pH was adjusted to 8–9 with 2 M hydrochloric acid, and the resulting precipitate was recovered by suction filtration and washed with water. In order to remove all traces of water, the compound was then dissolved in dichloromethane, dried with sodium sulfate and concentrated by evaporation. Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution.

Crystal data

 $\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}_2$
 $M_r = 252.27$
Monoclinic, $P2_1$
 $a = 6.5110$ (13) Å
 $b = 5.172$ (1) Å
 $c = 18.763$ (4) Å
 $\beta = 99.14$ (3)°
 $V = 623.8$ (2) Å³ $Z = 2$
 $D_x = 1.343$ Mg m⁻³
Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 293$ (2) K
Block, colourless
 $0.30 \times 0.10 \times 0.10$ mm

Data collection

Enraf–Nonius CAD-4
diffractometer
 $\omega/2\theta$ scans
Absorption correction: ψ scan
(North *et al.*, 1968)
 $T_{\min} = 0.973$, $T_{\max} = 0.991$
1483 measured reflections

1360 independent reflections
900 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.032$
 $\theta_{\text{max}} = 26.0^\circ$
3 standard reflections
every 200 reflections
intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.063$
 $wR(F^2) = 0.181$
 $S = 1.06$
1360 reflections
172 parameters

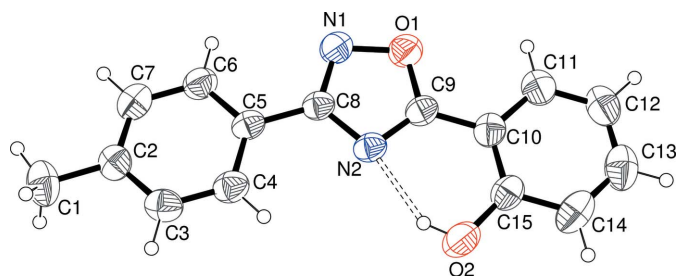
H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.099P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.21 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.21 \text{ e } \text{\AA}^{-3}$

Table 1Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O2-H2A\cdots N2$	0.82	1.94	2.663 (5)	147

Due to insignificant anomalous scattering, Friedel pairs were merged before refinement. All H atoms were positioned geometrically, with $N-H = 0.86 \text{ \AA}$ and $C-H = 0.93\text{--}0.96 \text{ \AA}$, and refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C,N})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97*

**Figure 1**

A view of the molecular structure of (I), showing 40% probability displacement ellipsoids (arbitrary spheres for the H atoms). The dashed lines indicate the hydrogen bond.

(Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *PLATON* (Spek, 2003).

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